

## Chemistry Chapter 3 Equations Worksheet 1

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### Chemistry Chapter 3 Equations Worksheet

3.zinc + hydrochloric acid ( zinc chloride + hydrogen. 4.iron + copper(II) sulfate ( iron (II) sulfate + copper. 5.copper + sulfuric acid ( copper (II) sulfate + hydrogen. 6.potassium dichromate + sulfur + water ( sulfur dioxide + potassium hydroxide + chromium (III) oxide.

### BALANCING CHEMICAL EQUATIONS - PRACTICE SHEET # 2

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The chemical equation also does not give any information about the speed of the reaction. Some chemical equations and reactions have diverse affect. Students likely find difficulty in balancing chemical equations worksheet. To help you resolve this issue, we have balancing equations worksheet with answers on our main website.

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### Chemistry Chapter 3 Equations Worksheet 1

Worksheet #3: Decomposition Reactions In decomposition reactions, one compound will break down into two or more parts. 1. barium carbonate Æ barium oxide + carbon dioxide BaCO<sub>3</sub> Æ BaO + CO<sub>2</sub> 2. magnesium carbonate Æ magnesium oxide + carbon dioxide MgCO<sub>3</sub> Æ MgO + CO<sub>2</sub> 3.

### Worksheet #3: Decomposition Reactions

Stoichiometry, part 3; Chapter 5 (Introduction to Solutions and Aqueous Reactions) Solution concentration; Dilutions; Making solutions; Molarity; Titrations; Molecular, Full, and Net-ionic equations; Chapter 6 (Gases) Gas laws; Gas stoichiometry; Chapter 7 (Thermochemistry) Calorimetry worksheet; Calorimetry exercises and answers

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### Chemistry Chapter 3 Equations Worksheet 1

Equation 11.3.2 is identical to Equation 11.3.1 except for the addition of abbreviations in parentheses to indicate the physical state of each species. The abbreviations are (s) for solid, (l) for liquid, (g) for gas, and (aq) for an aqueous solution, a solution of the substance in water.. Consistent with the law of conservation of mass, the numbers of each type of atom are the same on both ...

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